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Developing a New Water-Soluble Porphyrin as a Potential Photodynamic Cancer Therapy Agent

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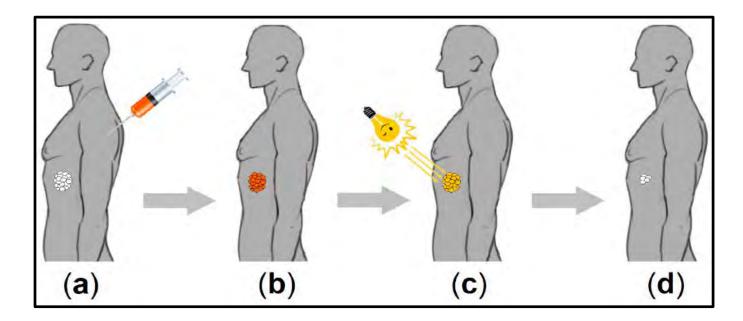
Developing a New Water-Soluble Porphyrin as a Potential Photodynamic Cancer Therapy Agent

A SENIOR THESIS BY CATHERINE L SHIRLEY

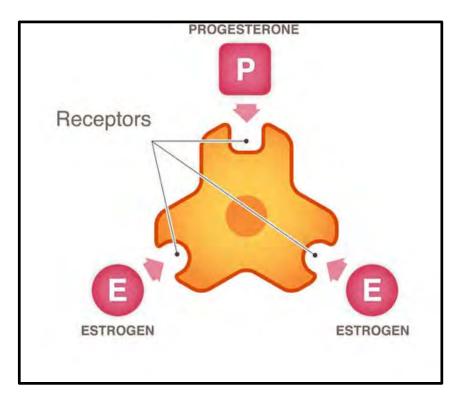
Background

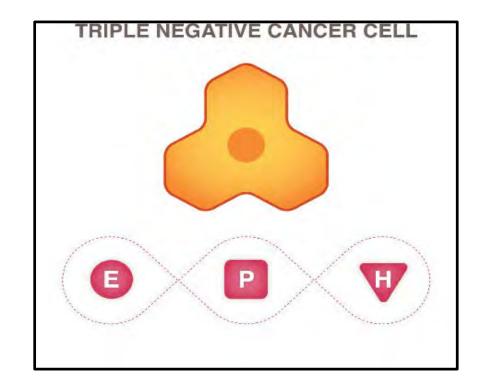
What is Photodynamic Therapy?

Photodynamic Therapy (PDT): treating cancer with light +a photosensitive compound



Why PDT?

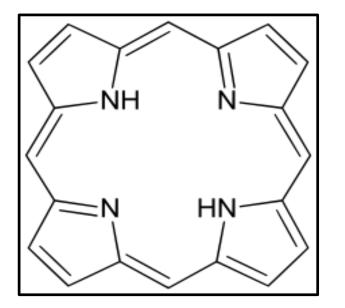




1

What are Porphyrins?

Ę



The structure of a porphyrin molecule. Chemical formula $C_{20}H_{14}N_4$

- ► Known for their role in hemoglobin
- Alternating single and double bonds give aromaticity and stabilization
- Create singlet, molecular oxygen, resulting in cytotoxicity of the tumor tissue cells

How is PDT Currently Being Used?

Three Photosensitizers

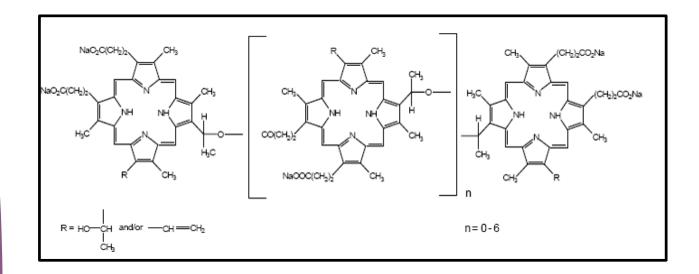


Foscan®

5-Aminolevulinic Acid

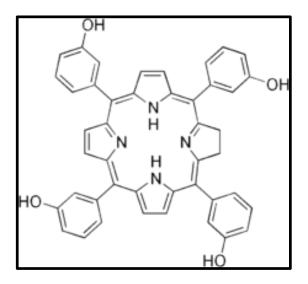


Photofrin®



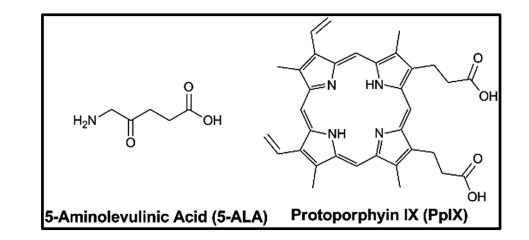
- Most studied photosensitive agent
- Reliable and easily activated
- Prolonged skin sensitivity

Foscan®



- Great tumor selectivity and deeper light penetration
- Only approved in Europe

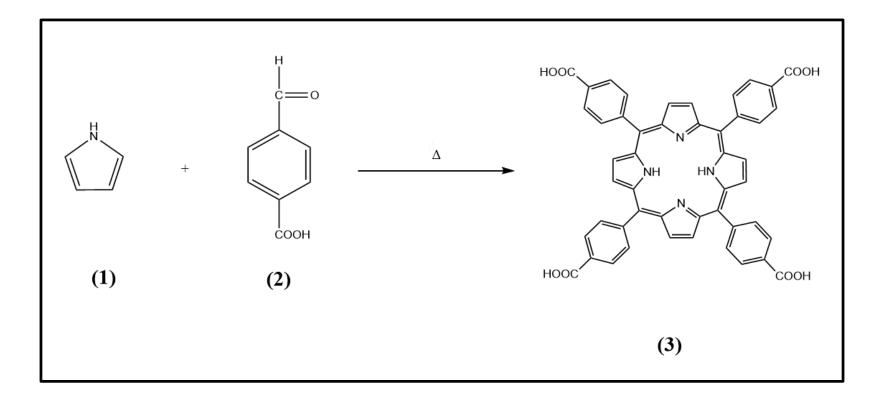
5-Aminolevulinic Acid



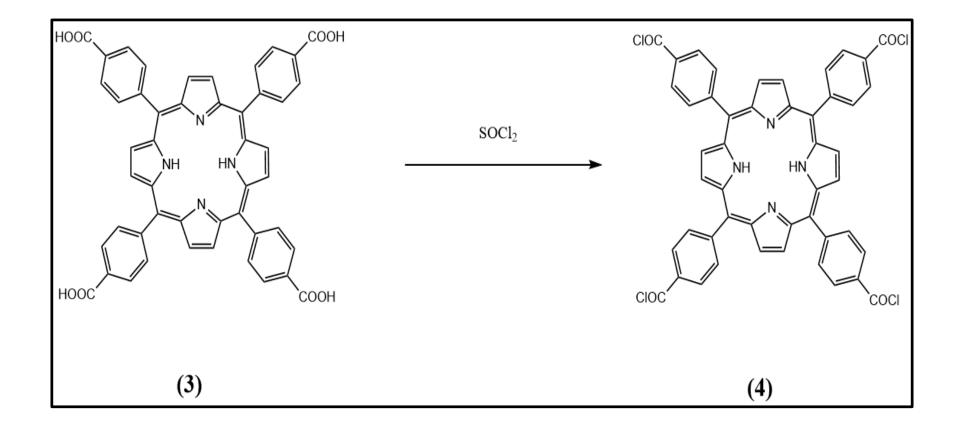
- Only for superficial lesions
- Clears from body in 48 hours
- Not effective for deep tumors

Methods: Porphyrin Synthesis

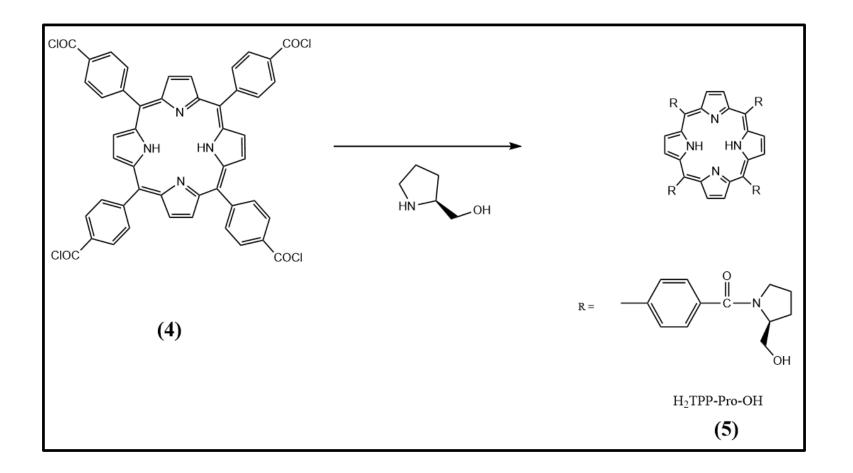
Formation of H₂TPPC



Formation of the Acid-Chloride Porphyrin



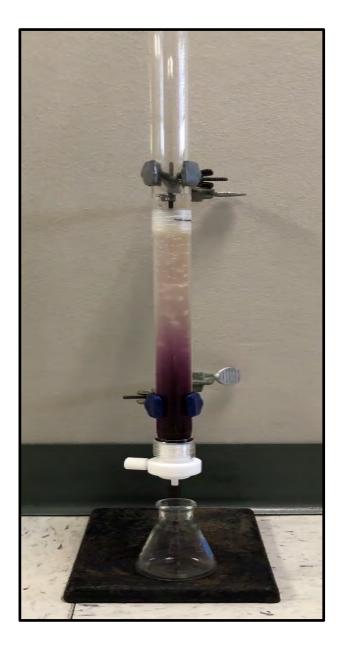
Formation of the Final Product, H₂TPP-Pro-OH



Purification

Two Methods to Purify

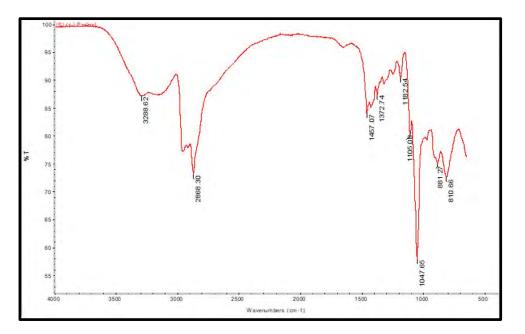
Sephadex LH-20 Sephadex G-50



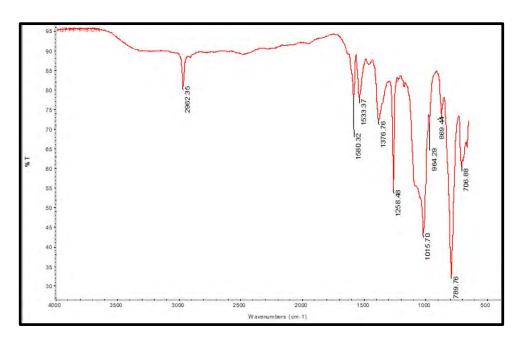
Characterization and Results

Infrared Spectroscopy

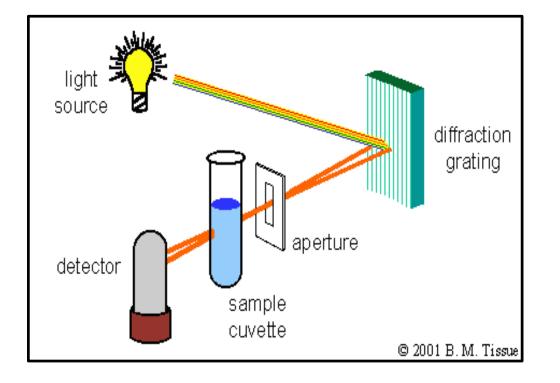
Infrared results for (S)-(+)-prolinol.



Infrared results for H₂TPP-Pro-OH.



Ultraviolet-Visible Spectroscopy



UV-Vis

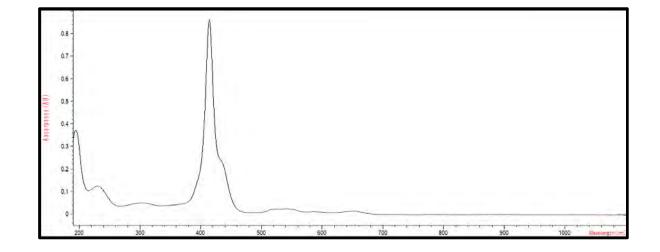
Spectroscopy diagram showing how light is transmitted



Ultraviolet-Visible Spectroscopy

Top: UV-Vis results for H₂TPP-Pro-OH.

Bottom: Peaks (nm) and epsilon $_{(\epsilon)}$ values for the UV-vis spectroscopy of H₂TPP-Pro-OH. Epsilon values calculated using Beer's Law: $A = \varepsilon c I$.

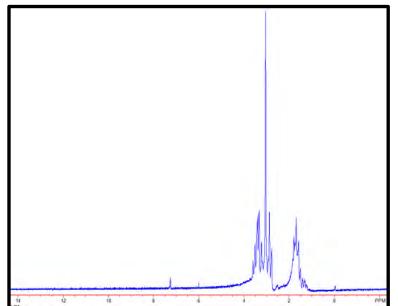


Peaks (nm)	Molar Absorptivity Coefficient, ε (cm ⁻¹ mM ⁻¹)
414	361
518	10.3
555	6.32
581	5.66
637	5.03

¹H Nuclear Magnetic Resonance Spectroscopy

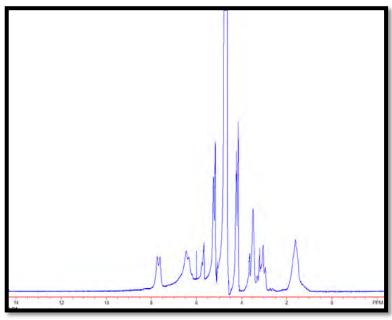
Results from ¹H NMR spectroscopy of the free amine,

(S)-(+)-prolinol, in CDCl₃.

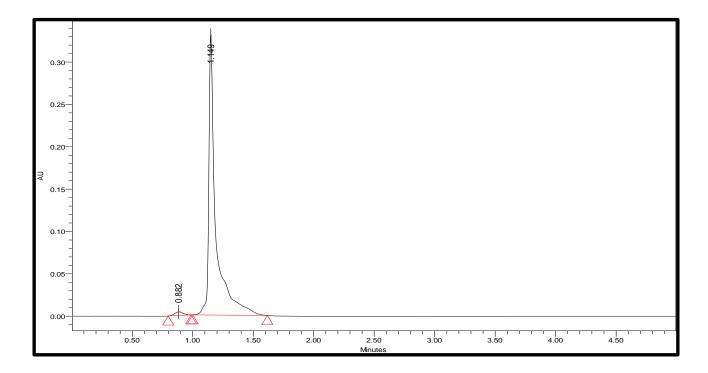


Results from ¹H NMR spectroscopy of final product,

 H_2 TPP-Pro-OH, in D_2 O.



High Performance Liquid Chromatography



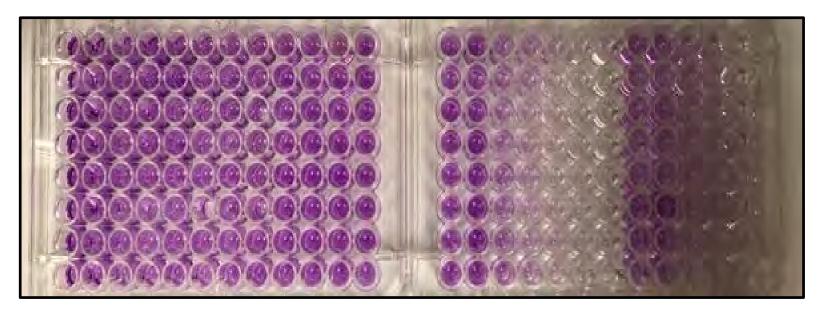
► HPLC results for H₂TPP-Pro-OH indicating 98% purity



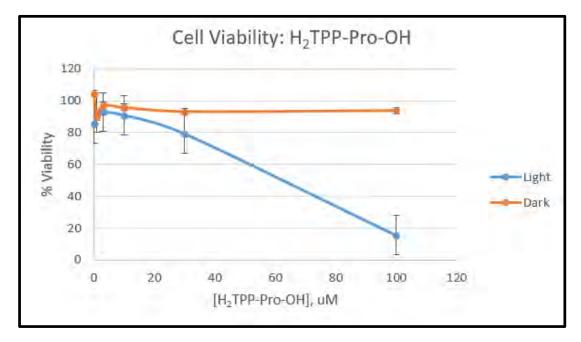


MTT Assay Procedure

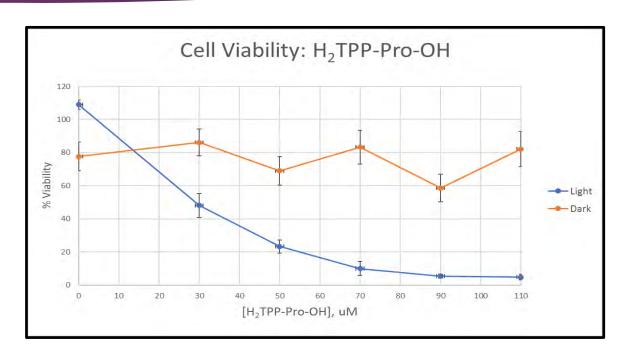
- Method of measuring cellular metabolic activity and determining cell viability
- MTT assay results after 72 hours of incubation. The left plate was entirely in dark conditions, while the right plate was exposed to light. The purple color indicates living cells.



MTT Assay Results



Spectrophotometric MTT assay results for trial one of the light and dark 96-well plates. The porphyrin concentrations used were 1, 3, 10, 30, and 100 µM



Spectrophotometric MTT assay results for trial two of the light and dark 96-well plates. The porphyrin concentrations used were 30, 50, 70, 90, and 110 µM

Conclusions

Future Work

Thank you!

DR. JOE BRADSHAW DR. TIM KNIGHT DR. TIM HAYES DR. TERRY CARTER TRAVIS HANKINS ALEX PODGUZOV



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